

This listing of claims replaces all prior listings:

1. (Currently Amended) A non-aqueous electrolyte secondary cell comprising:

(a) a cathode comprising  $\text{Li}_x\text{Fe}_y\text{PO}_4$  and having a particle diameter not greater than 1 micrometer and wherein  $0 < x \leq 2$  and  $1 \leq y \leq 2$ ;

(b) a binderless anode comprising:

(1) a molded body,

(2) a sintered mesophase carbon material prepared by sintering a mesophase carbon material after being molded into a body of the non-aqueous electrolyte secondary cell, said sintered mesophase carbon material being capable of doping/dedoping lithium, and

(3) an anode active material comprising Li and a tin or silicon, containing metal material which forms an alloy or a compound with Li; and

(c) a non-aqueous electrolyte solution.

2-3. (Canceled)

4. (Currently Amended) A non-aqueous electrolyte secondary cell comprising:

(a) a cathode having (1) a molded body, (2) a cathode active material, and (3) a conductive agent, said cathode active material comprising  $\text{Li}_x\text{Fe}_y\text{PO}_4$  and having a particle diameter not greater than 1 micrometer wherein  $0 < x \leq 2$  and  $1 \leq y \leq 2$ ;

(b) a binderless anode having (1) a molded body, (2) a sintered mesophase carbon material that is (i) sintered after being molded into a body of the non-aqueous electrolyte secondary cell and is (ii) capable of doping/dedoping lithium, and (3) an anode active material comprising Li and a tin or silicon metal material which forms an alloy or a compound with Li; and

(c) a non-aqueous electrolyte solution.

5-14. (Canceled)

15. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 1, wherein said non-aqueous electrolyte solution comprises an electrolyte salt and a non-aqueous solvent.

16. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 15, wherein said electrolyte salt is a lithium salt having ion conductivity.

17. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 16, wherein said lithium salt is selected from the group consisting of  $\text{LiClO}_4$ ,  $\text{LiAsF}_6$ ,  $\text{LiPF}_6$ ,  $\text{LiBF}_4$ ,  $\text{LiB}(\text{C}_6\text{H}_5)_4$ ,  $\text{LiCl}$ ,  $\text{LiBr}$ ,  $\text{CH}_3\text{SO}_3\text{Li}$ ,  $\text{N}(\text{C}_n\text{F}_{2n}\text{SO}_2)_2\text{Li}$ , and mixtures thereof.

18. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 15, wherein said non-aqueous solvent is selected from the group consisting of propylene carbonate, ethylene carbonate, 1,2-dimethoxyethane, 1,2-diethoxyethane, diethyl carbonate, methyl ethyl carbonate, dimethyl carbonate,  $\gamma$ -butyrolactone, tetrahydrofuran, 1,3-dioxolane, 4-methyl-1,3-dioxolane, diethyl ether, sulfolane, methyl sulfolane, acetonitrile, propionitrile, and mixtures thereof.

19-22. (Canceled)

23. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 4, wherein said non-aqueous electrolyte solution comprises an electrolyte salt and a non-aqueous solvent.

24. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 23, wherein said electrolyte salt is a lithium salt having ion conductivity.

25. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 24, wherein said lithium salt is selected from the group consisting of  $\text{LiClO}_4$ ,  $\text{LiAsF}_6$ ,  $\text{LiPF}_6$ ,  $\text{LiBF}_4$ ,  $\text{LiB}(\text{C}_6\text{H}_5)_4$ ,  $\text{LiCl}$ ,  $\text{LiBr}$ ,  $\text{CH}_3\text{SO}_3\text{Li}$ ,  $\text{N}(\text{C}_n\text{F}_{2n}\text{SO}_2)_2\text{Li}$ , and mixtures thereof.

26. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 23, wherein said non-aqueous solvent is selected from the group consisting of propylene carbonate, ethylene carbonate, 1,2-dimethoxyethane, 1,2-diethoxyethane, diethyl carbonate, methyl ethyl carbonate, dimethyl carbonate,  $\gamma$ -butyrolactone, tetrahydrofuran, 1,3-dioxolane, 4-methyl-1,3-dioxolane, diethyl ether, sulfolane, methyl sulfolane, acetonitrile, propionitrile, and mixtures thereof.

27. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 1, wherein the cathode further comprises a conductive material and a binder.

28. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 1, wherein the anode further includes a molded and sintered current collector material combined with said sintered carbon material.

29. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 1, wherein the tin or silicon containing metal material includes a metal selected from the list of elements consisting of B, Mg, Ti, Mo, Co, Ni, Ca, Cr, Cu, Fe, Mn, Nb, Ta, V, and W.

30. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 4, wherein said cathode further comprises a conductive material and a binder.

31. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 4, wherein the anode further includes a molded and sintered current collector material combined with said sintered carbon material.

32. (Cancelled)

33. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 1, wherein the molded body of the anode has an outer diameter of 15.6 mm, a height of 0.8 mm, and a weight of 180 mg.

34. (Previously Presented) The non-aqueous electrolyte secondary cell of Claim 4, wherein the molded body of the anode has an outer diameter of 15.6 mm, a height of 0.8 mm, and a weight of 180 mg.